

Minutissamides A–D, Antiproliferative Cyclic Decapeptides from the Cultured Cyanobacterium *Anabaena minutissima*

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ABSTRACT:



Four cyclic decapeptides, minutissamides A–D (1–4), were isolated from the cultured cyanobacterium Anabaena minutissima (UTEX 1613). The planar structures were determined using various spectroscopic techniques including HRESIMS and 1D and 2D NMR experiments. The absolute configurations of the α -amino acid residues were assigned using Marfey's method after acid hydrolysis. The absolute configuration of a β -amino acid residue was assigned by a combination of the advanced Marfey's method, *J*-based configurational analysis, and ROE spectroscopic analysis. The structures of minutissamides A–D (1–4) were characterized by the presence of three nonstandard α -amino acid residues (two α , β -dehydro- α -aminobutyric acids and one *N*-methylated Asn) and one β -amino acid residue (2-hydroxy-3-amino-4-methyldodecanoic acid or 2-hydroxy-3-amino-4-methylhexadecanoic acid). Minutissamides A–D (1–4) exhibited antiproliferative activity against the HT-29 human colon cancer cell line with IC₅₀ values of 2.0, 20.0, 11.8, and 22.7 μ M, respectively.

yanobacteria (blue-green algae) have been shown to be prolific producers of bioactive secondary metabolites.^{1–3} A major class of secondary metabolites from cyanobacteria is polyketide-conjugated nonribosomal peptides, commonly known as lipopeptides.^{4,5} Numerous cyclic lipopeptides have been isolated from both freshwater and marine cyanobacteria with ring sizes of up to 12 amino acid residues. Cyclic deca- and undecapeptides, containing 10 and 11 residues, have been reported to exhibit a wide spectrum of biological activities, including antifungal,^{6,7} cardiotonic,^{8,9} and cytotoxic.^{10,11} A common structural feature of cyclic deca- and undecapeptides is the occurrence of one lipophilic β -amino acid residue with a structurally unique side chain.¹² Amination of fatty acids at C-3 forming β -amino acids provides the branching point via an amide bond with other α amino acids, whereas C-2 is usually a methylene, methylated or hydroxylated. Further fatty acid modifications include methylation, oxidation, chlorination, and/or hydroxylation of the side chain, resulting in the high diversity of modified β -amino fatty acids. Most of the cyclic lipopeptides in this class also possess nonstandard amino acids, such as N-methylated α -amino acids,

 α -hydroxy or β -hydroxy amino acids, and Dhb (α , β -dehydro- α -aminobutyric acid).

In our continuing search for secondary metabolites from laboratory-cultured cyanobacteria, we evaluated the cell extract of cultured Anabaena minutissima (UTEX 1613). Herein we report the isolation, structure elucidation, and biological activity of four new cyclic decapeptides, named minutissamides A–D, each possessing two Dhbs, one N-methylated Asn, and one β -amino acid, Hamd (2-hydroxy-3-amino-4-methyldodecanoic acid) or modified Hamh (2-hydroxy-3-amino-4-methyldodecanoic acid). The planar structures were determined using a combination of spectroscopic techniques including HRESIMS, tandem MS, and 1D and 2D NMR (COSY, TOCSY, HSQC, HMBC, and ROESY) experiments. The absolute configurations of the α -amino acid residues were determined by Marfey's method, whereas the stereoconfiguration of the Hamd residue

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RESULTS AND DISCUSSION

A. minutissima (UTEX 1613) was grown in Z media.¹³ The freeze-dried cells were extracted with a mixture of CH_2Cl_2 and MeOH (1:1) and dried in vacuo. The cell extract was subsequently fractionated by column chromatography using Diaion HP-20 resin and an increasing amount of isopropyl alcohol (IPA) in H₂O. LC-MS analysis of the third fraction (40% IPA) indicated the presence of peptides with molecular weights of 1118, 1152, 1188, and 1190 Da. Subsequent Sephadex LH-20 column chromatography followed by reversed-phased HPLC yielded minutissamides A–D (1–4).

Minutissamide A (1) was obtained as a colorless, amorphous powder. The molecular formula of 1 was determined as C51H83N13O15 by HRESIMS analysis $(m/z \ 1140.6049 \ [M + Na]^+)$. The ¹H NMR spectrum in DMSO- d_6 showed a signal distribution typical for lipopeptides, including several exchangeable amide NH signals (6.5–10.5 ppm), signals in the α -proton region (4–6 ppm), methylene signals (1.2-1.3 ppm), and numerous aliphatic doublet and triplet methyl signals (0.5–2.0 ppm). The structures of the amino acid residues were elucidated by interpretation of the COSY and TOCSY spectra (Figure 1 and Table 1) and indicated the presence of 10 residues, including nine α -amino acid residues and one β -amino acid residue. Six of the α -amino acid residues were identified as the standard amino acids proline (Pro), threonine (Thr), alanine (Ala), valine (Val), and two asparagines (Asn). In addition, the structures of three nonstandard amino acid residues were determined by analysis of COSY, HSQC, and HMBC data (Figure 1). An HMBC correlation from the N-methyl singlet ($\delta_{\rm H}$ 2.96) to the C-2 $_{\rm NMeAsn}$ ($\delta_{\rm C}$ 49.4) in combination with a COSY correlation between NMeAsn H-2 $(\delta_{\rm H} 5.53)$ and NMeAsn H₂-3 $(\delta_{\rm H} 2.00$ and 3.02) identified the structure of N-methylasparagine (NMeAsn). The presence of two Dhb residues was deduced by COSY correlations observed between the methyl protons (H₃-4_{Dhb1}, $\delta_{\rm H}$ 1.90 and H₃-4_{Dhb2}, $\delta_{\rm H}$ 1.66) and the corresponding olefinic proton (H-3_{Dhb1}, $\delta_{\rm H}$ 5.71 and H-3_{Dhb2}, $\delta_{\rm H}$ 5.30), and HMBC correlations from the methyl protons $(H_3-4_{Dhb1} \text{ and } H_3-4_{Dhb2})$ to the corresponding olefinic carbon (C-2_{Dhb1}, $\delta_{\rm C}$ 130.1 and C-2_{Dhb2}, $\delta_{\rm C}$ 133.0). The 2-hydroxy-3-amino-4-methyl portion in Hamd was identified by sequential COSY correlations between H-2 ($\delta_{\rm H}$ 4.18)/H-3 ($\delta_{\rm H}$ 3.93)/H-4 ($\delta_{
m H}$ 1.65), as well as COSY correlations between H-3



Figure 1. Key 2D correlations of minutissamide A (1) that were used for the determination of the planar structure of 1.

 $(\delta_{\rm H}~3.93)$ and 3-NH $(\delta_{\rm H}~6.85)$ and between H-4 $(\delta_{\rm H}~1.65)$ and 4-Me $(\delta_{\rm H}~0.58).$ Further COSY correlations from H-4 $(\delta_{\rm H}~1.65)$ to H₂-5 $(\delta_{\rm H}~1.17$ and 1.61), H₂-5 to H₂-6 $(\delta_{\rm H}~1.18$ and 1.24), and H₂-6 to the region of highly overlapped methylene signals (H₂-7–11, $\delta_{\rm H}~1.26-1.27$), corresponding to five carbons in the HSQC spectrum ($\delta_{\rm C}~30.1, 29.2, 29.7, 31.8,$ and 22.6), and finally to the methyl triplet protons H₃-12 $(\delta_{\rm H}~0.87)$, completed the structure of Hamd.

The sequence of the 10 residues was established by analysis of the ROESY and selective HMBC spectra (Figure 1 and Table 1). A selective HMBC spectrum for the carbonyl region was used to resolve the overlapped amide carbonyl signals.¹⁴ HMBC correlations from Dhb₂-NH ($\delta_{\rm H}$ 9.17) to Val C-1 ($\delta_{\rm C}$ 169.2), from Val H-2 ($\delta_{\rm H}$ 4.29) to Hamd C-1 ($\delta_{\rm C}$ 170.1), and from Hamd H-3 $(\delta_{\rm H} 3.93)$ to Pro C-1 $(\delta_{\rm C} 171.9)$ suggested a partial sequence of Dhb₂-Val-Hamd-Pro. ROE correlations between Dhb₂-NH ($\delta_{
m H}$ 9.17) and Val H-2 ($\delta_{\rm H}$ 4.29), between Val-NH ($\delta_{\rm H}$ 6.85) and Hamd H-2 ($\delta_{\rm H}$ 4.18), and between Hamd-NH ($\delta_{\rm H}$ 6.85) and Pro H-2 ($\delta_{\rm H}$ 4.19) confirmed this partial sequence. HMBC correlations from Pro H-2 ($\delta_{\rm H}$ 4.19) to NMeAsn C-1 ($\delta_{\rm C}$ 168.3) and from NMeAsn N-Me ($\delta_{\rm H}$ 2.96) to Thr C-1 ($\delta_{\rm C}$ 170.4), together with ROE correlations between Pro H_2-5 ($\delta_{
m H}$ 4.10) and NMeAsn H-2 ($\delta_{\rm H}$ 5.53) and between NMeAsn N-Me ($\delta_{\rm H}$ 2.96) and Thr H-2 ($\delta_{\rm H}$ 4.63), further expanded the sequence to Dhb₂-Val-Hamd-Pro-NMeAsn-Thr. HMBC correlations from Thr NH $(\delta_{
m H}$ 7.30) to Ala C-1 $(\delta_{
m C}$ 172.5) and from Ala NH $(\delta_{
m H}$ 8.27) to Asn1 C-1 ($\delta_{\rm C}$ 171.2), together with ROE correlations between Thr-NH $(\delta_{
m H}$ 7.30) and Ala H-2 $(\delta_{
m H}$ 4.29) and between Ala-NH $(\delta_{\rm H} 8.27)$ and Asn₁ H-2 $(\delta_{\rm H} 4.40)$, established the sequence of Dhb2-Val-Hamd-Pro-NMeAsn-Thr-Ala-Asn1, leaving two residues (Dhb₁ and Asn₂) unassigned. The complete sequence of Dhb₂-Val-Hamd-Pro-*N*MeAsn-Thr-Ala-Asn₁-Dhb₁-Asn₂ was deduced by an HMBC correlation from Asn₁-NH ($\delta_{\rm H}$ 8.45) to Dhb₁ C-1 ($\delta_{\rm C}$ 163.4) and a ROE correlation between Dhb₁-NH $(\delta_{\rm H} 10.18)$ and Asn₂ H-2 $(\delta_{\rm H} 4.92)$. Lastly, HMBC correlations from Asn₂ H-2 ($\delta_{\rm H}$ 4.92) and NH ($\delta_{\rm H}$ 8.73) to Dhb₂ C-1 ($\delta_{\rm C}$ 163.6) closed the ring, completing the planar structure of 1.

Mass fragmentation analysis was performed to further verify the structure of 1.¹⁵ The ESIMS/MS analysis, using a quadrupole-time-of-flight (Q-TOF) tandem mass spectrometer, supported the proposed planar structure of 1. The parent peak was observed at m/z [M + H]⁺ 1118.6. The ring-opening occurred

		minutissamide A (1)						minutissamide B $(2)^a$		
		$\delta_{c}{}^{b}$	${\delta_{\mathrm{H}}}^{c}$	mult. (J in Hz)	COSY	HMBC	ROESY	δ_{c}^{f}	$\delta_{ m H}{}^{c}$	mult.(<i>J</i> in Hz)
Hamd or Hamcd	1	170.1						170.0		
	2	70.0	4.18	d (4.8)	3	3		70.0	4.18	overlapped
	3	56.6	3.93	m	2, 4	$1,^{e} 2, 1_{Pro}^{e}$		56.6	3.92	М
	4	32.6	1.65	m	3, 5, 4-Me			32.6	1.65	m
	5	33.8	1.17	m	4, 6			33.7	1.17	m
			1.61	m					1.61	m
	6	25.9	1.18	m	5, 6			25.9	1.16	m
			1.24	m					1.23	m
	7	30.1	1.26	m	6, 8			30.1	1.23	m
	8	29.2	1.26	m	7, 9			29.7	1.23	m
	9	29.7	1.26	m	8, 10			29.1	1.26	m
	10	31.8	1.26	m	10, 11			26.6	1.37	m
	11	22.6	1.27	m	11, 12			32.4	1.70	m
	12	14.5	0.87	t (7.2)	11	10, 11		45.8	3.62	t (6.6)
	2-OH		5.51	br s	2					nd ^d
	3-NH		6.85	partly overlapped	3	$1_{\rm Pro}^{e}$	3, 2 _{Pro}		6.88	partly overlapped
	4-Me	16.3	0.58	d (6.6)	4	3, 4, 5	3	16.4	0.57	d (6.6)
Pro	1	171.9						171.9		
	2	60.5	4.19	m	3	3, 4, 1 _{NMeAsn}		60.4	4.18	m
	3	30.7	1.91	m	2, 4	1		30.7	1.92	m
	4	24.2	1.85	m	3, 5			24.2	1.83	m
			1.92	m					1.92	m
	5	47.3	3.26	m	4	2	2 _{NMeAsn}	47.3	3.25	m
			4.10	m					4.09	m
NMeAsn	1	168.3						168.3		
	2	49.4	5.53	dd (12.0, 3.0)	3	$1,^{e} 3, 4^{e}$	5 _{Pro} , 3, N-Me	49.3	5.53	br d (12.0)
	3	34.4	2.00	m	2, 4	$1^{e}, 2^{e}, 4^{e}$		34.4	2.00	m
			3.02	dd (15.6, 12.0)					3.01	m
	4	172.1						172.1		
	N-Me	30.7	2.96	S		2, 1_{Thr}^{e}	2_{Thr} 3_{Thr}	30.7	2.95	S
	NH_2		5.92	S		3			6.03	m
			7.52	S					7.52	m
Thr	1	170.4						170.4		
	2	55.3	4.63	dd (8.4, 3.6)	2-NH, 3	$1, 3, 1_{Ala}$	N-Me, 3, 4	55.3	4.62	m
	3	66.8	3.92	m	2, 4			66.8	3.92	m
	4	20.0	1.00	d (6.0)	3	2, 3		19.9	0.99	d (6.6)
	3-0H		5.13	br s $1(7.0)$	2		2.2		7.20	nd"
41-	NH	172.5	/.30	d (7.8)	2	1_{Ala}	2, 3	172.5	7.30	a (8.4)
Ala	1	1/2.5	4.20	- (72)	2 111 2	101 0		1/2.5	4 2 0	
	2	48.9	4.29	q(7.2)	2-NH, 3	1, 1_{Asn1}		48./	4.28	m 1 (7.2)
	3 NH	17.5	1.23	d(7.2)	2	1, 2		17.5	1.22	d (7.2)
A cm 1	1	171.2	0.27	u (8.4)	Z	2, 1 _{Asn1}	L_{1} L_{Asn1} , Mn_{Thr} , Mn_{Asn1}	171.2		
Asn1	2	50.5	4.40		2 NH 2	2	2	50.2	4 20	
	2	30.5	4.40	III m	2-INFL, 5	3 1 ^e 2 4^{e}	3	30.5	4.59	lii m
	3	30.2	2.30	dd(186.72)	2	1, 2, 4		30.2	2.33	dd(168.72)
	4	172.2	2.07	du (18.0, 7.2)				172.2	2.00	uu (10.8, 7.2)
	т NH	1/3.4	815	d(72)	2	1.50.0	2	1/3.4	815	d(72)
	NH.		7.04	s (7.2)	2	*Dhbl	-		7.04	s (7.2)
	11112		7 58	s		5			7 59	s
Dhb1	1	163.4	/.50	U U				163.4		-
	2	130.1						130.1		
	-	100.1						100.1		

Table 1. NMR Spectroscopic Data for Minutissamides A and B (1 and 2) in DMSO- d_6

Table 1. Continued

		minutissamide A (1)					minutissamide B $(2)^a$			
		$\delta_{c}{}^{b}$	$\delta_{ ext{H}}{}^{c}$	mult. (J in Hz)	COSY	HMBC	ROESY	δ_{c}^{f}	${\delta_{\mathrm{H}}}^{c}$	mult.(<i>J</i> in Hz)
	3	125.6	5.71	q (7.2)	4	1, 2, 4	NH	125.6	5.71	q (7.2)
	4	13.7	1.90	d (7.2)	3	2, 3		13.6	1.89	d (7.2)
	NH		10.18	s		1, 1 _{Asn2} ^e	3, 2 _{Asn2}		10.20	S
Asn2	1	172.4						172.4		
	2	49.4	4.92	q (7.2)	2-NH, 3	$1,^{e} 3, 1_{\text{Dhb2}}^{e}$	3	49.4	4.91	q (7.8)
	3	39.1	2.67	dd (15.0, 7.2)	2	1, ^e 2, 4 ^e		39.2	2.67	dd (15.0, 7.2)
			2.77	dd (15.0, 7.8)					2.77	dd (15.0, 7.8)
	4	172.6						172.7		
	NH		8.73	d (7.8)	2	$1_{\rm Dhb2}$			8.74	d (7.8)
	NH_2		7.11	S		3			7.12	S
			7.58	s					7.59	S
Dhb2	1	163.6						163.6		
	2	133.0						133.0		
	3	116.5	5.30	q (7.2)	4	1, 2, 4	NH	116.4	5.29	q (7.2)
	4	13.5	1.66	d (7.2)	3	2, 3		13.3	1.65	d (7.2)
	NH		9.17	s		$1,^{e}3, 1_{Val}^{e}$	3, 2 _{Val}		9.21	S
Val	1	169.2						169.2		
	2	56.2	4.29	d (10.8)	2-NH, 3	$1,^e 1_{\mathrm{Hamd}}^e$		56.2	4.28	m
	3	33.1	1.75	m	2, 4	1^e		33.1	1.74	m
	4	19.4	0.83	d (6.6)	3	2, 3, 4'		19.3	0.82	d (6.6)
	4′	19.1	0.90	d (6.6)	3	2, 3, 4		19.0	0.89	d (6.6)
	NH		6.85	partly overlapped	2		2 _{Hamd}		6.88	partly overlapped

^{*a*} Complete NMR table for minutissamide B (2) including 2D NMR data can be found in the Supporting Information. ^{*b*} Carbon chemical shifts were assigned from DEPT-Q spectrum measured at 226 MHz. ^{*c*} Measured at 600 MHz. ^{*d*} nd: not detected. ^{*e*} Assigned from the selective HMBC spectrum. ^{*f*} Assigned from the HSQC and HMBC spectra.

between Pro and NMeAsn, forming an acylium ion. The continuous fragmentation yielded the fragment ions at m/z 990.6 $[M + H - NMeAsn]^+$, 889.5 $[M + H - NMeAsn - Thr]^+$, 704.4 $[M + H - NMeAsn - Thr - Ala - Asn_1]^+$, 621.4 $[M + H - NMeAsn - Thr - Ala - Asn_1 - Dhb_1]^+$, 507.4 $[M + H - NMeAsn - Thr - Ala - Asn_1 - Dhb_1 - Asn_2]^+$, and 396.3 $[M + H - NMeAsn - Thr - Ala - Asn_1 - Dhb_1 - Asn_2 - Dhb_2]^+$ (Figure 2). This fragmentation pattern was in complete agreement with the structure determined by 2D NMR analysis.

Minutissamide A (1) contained three different types of stereogenic elements, the geometric configuration of Dhbs, the configurations of the α -amino acid residues, and the configurations of the three consecutive asymmetric centers in the Hamd residue. A number of techniques were used for the assignment of stereoconfiguration of 1, including Marfey's method and the advanced Marfey's method, J-based configurational analysis, and ROE correlations. A strong ROE correlation between the respective NH ($\delta_{\rm H}$ 10.18 or $\delta_{\rm H}$ 9.17) and the corresponding olefinic proton (H-3, $\delta_{\rm H}$ 5.71 or $\delta_{\rm H}$ 5.30) was observed for both Dhb₁ and Dhb₂, assigning the geometric configuration of both double bonds as E (Figure 1). The absolute configurations of the common amino acids and NMeAsn were assigned as L-Pro, L-Thr, L-Asn1, L-Asn2, L-Val, D-Ala, and L-NMeAsn by chromatographic comparison of Marfey's derivatives of the acid hydrolysate of 1 and appropriate amino acid standards.¹⁶ The relative configurations of the three asymmetric centers (C-2, C-3, and C-4) in the Hamd residue were established by a combination of *J*-based configurational analysis and ROE correlations (Figure 3). Homonuclear $({}^{3}J_{H-H})$ coupling constants were acquired from

the DQF-COSY spectrum, and the phase-sensitive HMBC spectrum was used to obtain the heteronuclear $({}^{2}J_{CH} \text{ and } {}^{3}J_{CH})$ coupling constants.¹⁷ The small ${}^{3}J_{\rm HH}$ (5.6 Hz) between H-2 and H-3, the large ${}^{3}J_{CH}$ (5.4 Hz) between H-3 and C-1, and the large $^{2}J_{CH}$ (-5.1 Hz) between H-3 and C-2 allowed for two possible conformations (A2 and A6) out of the six conformations shown in Figure 3a. The conformation A6 was found to be a correct conformation based on a ROE correlation observed between H-2 and H-4, assigning the relative configuration between C-2 and C-3 as "erythro". The same method was applied for the assignment of the relative configuration between C-3 and C-4. On the basis of the large ${}^{3}J_{HH}$ (11.0 Hz) measured between H-3 and H-4, only two conformations (B3 and B4) were possible, as shown in Figure 3b. A ROE correlation observed between H-2 and H₂-5 confirmed B3 as the correct confirmation, assigning the relative configuration between C-3 and C-4 as "threo". The nearly identical carbon chemical shifts of C-1, C-2, C-3, and C-4 of Hamd ($\delta_{\rm C}$ 170.1, 70.0, 56.6, and 32.6) to those of Hamh ($\delta_{\rm C}$ 169.8, 69.9, 56.2, and 32.4) in puwainaphycin E provided further evidence for this relative configuration.¹⁰ Because the relative configurations of C2-C3 and C3-C4 were erythro and three, respectively, the absolute configurations of the three stereogenic centers (C-2, C-3, and C-4) in the Hamd residue could be either "RRS" or "SSR". The advanced Marfey's method was applied to assign the absolute configuration of the Hamd residue.^{18,19} The acid hydrolysate of 1 was derivatized with L- and DL-FDLA, respectively. Subsequent LC-MS analysis of DL-FDLA derivatives identified two peaks of the FDLA derivatives of Hamd $(m/z 538 [M - H]^{-})$ at 43.3 and 48.8 min, while the L-FDLA



Figure 2. ESIMS/MS fragmentation of minutissamide A (1).



Figure 3. Newman projections for (a) C-2/C-3 and (b) C-3/C-4. All possible relative conformations are shown. The DQF-COSY and phase-sensitive HMBC spectra were used for the calculation of homo- and heteronuclear coupling constants. Labels below projections denote the predicted size of coupling constants. The predicted values that are consistent with observed values are highlighted by a box. Observed ROE correlations are presented as double-headed arrows: (a) H-2/H-4 and (b) H-2/H₂-5.

derivative of Hamd resulted in one peak at 48.8 min. Previous application of this method for the determination of the absolute configuration of Ahda (amino-2-hydroxydecanoic acid) identified that the *R*-L and *S*-D derivatives are more hydrophobic than the *R*-D and *S*-L derivatives.²⁰ In our experiment, the L-FDLA derivative of 1 corresponded to the later peak of DL-FDLA derivatives, suggesting the presence of *R*-L. As a result, the absolute configuration of C-3 in Hamd was assigned as *R*, which allowed for the assignment of the absolute configurations at C-2, C-3, and C-4 in Hamd as "*RRS*".

Minutissamide B (2) was obtained as a colorless, amorphous powder. The molecular formula of 2 was deduced as $C_{51}H_{82}ClN_{13}O_{15}$ on the basis of HRESIMS analysis (m/z 1174.5680 [M + Na]⁺). A M + 2 isotope peak verified the presence of chlorine in 2. The ¹H and ¹³C NMR spectra of 2 were almost identical to those of 1 except for the Hamd residue, where the terminal methyl proton signal had been replaced by a new triplet methylene resonance at 3.62 ppm. This indicated that chlorine was positioned on the terminal carbon of Hamd, thus renaming it Hamcd (2-hydroxy-3-amino-4-methyl-12chlorododecanoic acid). The sequential COSY correlation network of

Table 2. ¹H and ¹³C NMR Data for Minutissamides C and D (3 and 4) in DMSO-d₆

		minutissamide C (3) ^a		minutissamide D $(4)^a$			
		$\delta_{\rm C}{}^b$	$\delta_{ m H}{}^c$	mult. (J in Hz)	δ_{c}^{b}	${\delta_{\mathrm{H}}}^{c}$	mult. (J in Hz)
Hamoh or Hamhh	1	170.0			170.1		
	2	70.0	4.17	d (4.8)	70.0	4.18	d (4.2)
	3	56.5	3.92	m	56.6	3.92	m
	4	32.6	1.65	m	32.6	1.65	m
	5	33.8	1.17	m	33.9	1.17	m
			1.62	m		1.61	m
	6	25.9	1.16	m	25.9	1.16	m
			1.23	m		1.23	m
	7	30.1	1.23	m	30.1	1.25	m
	8	29.1	1.23	m	29.1	1.25	m
	9	29.7	1.23	m	29.7	1.25	m
	10	29.4	1.23	m	29.4	1.25	m
	11	29.4	1.23	m	29.4	1.25	m
	12	23.6	1.44	q (7.2)	25.8	1.16	m
						1.35	m
	13	41.9	2.38	t (7.2)	35.4	1.26	m
						1.32	m
	14	211.5			71.5	3.28	m
	15	35.4	2.40	q (7.8)	30.1	1.16	m
						1.35	m
	16	8.2	0.90	t (7.8)	10.6	0.83	t (7.2)
	2-OH			nd^d			nd^d
	3-NH		6.84	partly overlapped		6.87	partly overlapped
	4-CH ₃	16.4	0.57	d (6.6)	16.5	0.57	d (6.6)
Pro	1	171.9			172.0		
	2	60.6	4.18	m	60.6	4.18	m
	3	30.8	1.92	m	30.8	1.92	m
	4	24.2	1.84	m	24.3	1.83	m
			1.91	m		1.91	m
	5	47.3	3.26	m	47.3	3.25	m
			4.10	m		4.09	m
NMeAsn	1	168.2			168.3		
	2	49.4	5.52	dd (12.0, 3.0)	49.1	5.53	dd (12.0, 3.0)
	3	34.4	2.00	m	34.4	1.99	m
			3.01	dd (16.2, 12.6)		3.01	dd (15.6, 12.6)
	4	172.1			172.1		
	N-Me	30.8	2.95	S	30.8	2.95	S
	NH_2		5.95	S		6.01	S
			7.52	S		7.52	S
Thr	1	170.4			170.4		
	2	55.3	4.62	dd (9.0, 3.6)	55.3	4.62	dd (8.4, 3.6)
	3	66.8	3.93	m	66.8	3.91	m
	4	20.0	0.99	d (6.0)	20.0	0.98	d (7.8)
	3-OH			nd^d			nd^d
	NH		7.29	d (9.0)		7.30	d (7.8)
Ala	1	172.6			172.6		
	2	48.9	4.28	m	49.0	4.28	m
	3	17.3	1.22	d (7.2)	17.4	1.22	d (7.2)
	NH		8.26	d (8.4)		8.26	d (8.4)
Asn1	1	171.2			171.2		
	2	50.5	4.39	td (7.2, 3.0)	50.5	4.39	td (7.2, 3.0)
	3	36.2	2.53	m	36.2	2.52	m

Table 2. Continued

			minutissamide C $(3)^a$			minutissamide D $(4)^a$			
		$\delta_{\rm C}{}^b$	$\delta_{ m H}{}^{c}$	mult. (J in Hz)	$\delta_{\rm C}{}^b$	$\delta_{ m H}{}^{c}$	mult. (J in Hz)		
			2.87	dd (16.8, 7.2)		2.87	dd (16.8, 7.2)		
	4	173.2			173.2				
	NH		8.45	d (7.2)		8.44	d (7.2)		
	NH ₂		7.04	S		7.04	S		
			7.58	S		7.59	S		
Dhb1	1	163.4			163.4				
	2	130.1			130.1				
	3	125.6	5.70	q (7.2)	125.8	5.70	q (7.8)		
	4	13.7	1.89	d (7.2)	13.7	1.89	d (7.8)		
	NH		10.18	S		10.19	S		
Asn2	1	172.4			172.4				
	2	49.4	4.90	q (7.2)	49.5	4.91	q (7.2)		
	3	39.0	2.66	dd (15.0, 7.2)	39.2	2.67	dd (15.0, 7.2)		
			2.76	dd (15.0, 7.8)		2.76	dd (15.0, 7.8)		
	4	172.6			172.7				
	NH		8.73	d (7.8)		8.73	d (7.8)		
	NH ₂		7.12	S		7.12	S		
			7.59	S		7.59	S		
Dhb2	1	163.6			163.7				
	2	133.0			133.0				
	3	116.4	5.29	q (7.2)	116.5	5.29	q (7.2)		
	4	13.5	1.65	d (7.2)	13.5	1.64	d (7.2)		
	NH		9.18	S		9.19	S		
Val	1	169.2			169.3				
	2	56.2	4.28	m	56.3	4.28	m		
	3	33.1	1.74	m	33.1	1.74	m		
	4	19.4	0.82	d (6.6)	19.4	0.82	d (6.6)		
	4′	19.1	0.89	d (6.6)	19.1	0.89	d (6.6)		
	NH		6.84	partly overlapped		6.86	partly overlapped		
¹ Complete NMR	tables for minutissan	nides $C(3)$ and	D (4) including	g 2D NMR data can be fou	ind in the Supp	orting Informat	ion. ^b Carbon chemical		

shifts were assigned from the DEPT-Q spectrum measured at 226 MHz. ^{*c*} Measured at 600 MHz. ^{*d*} nd: not detected.

H₂-12/H₂-11/H₂-10 and the slightly downfield chemical shifts of H₂-11 ($\delta_{\rm H}$ 1.70) and H₂-10 ($\delta_{\rm H}$ 1.37) further supported the structure of **2**. The stereoconfiguration of **2** was determined by comparison of proton and carbon chemical shifts as well as specific rotation to those of **1**. The carbon chemical shifts of all of the stereogenic centers in **2** were similar to those observed for **1** (deviation <0.2 ppm), suggesting the same relative configurations. The specific rotations of both **1** and **2** displayed positive values (**1**, $[\alpha]_{\rm D}$ +4; **2**, $[\alpha]_{\rm D}$ +5). On the basis of these data and a shared biosynthesis, we propose that the absolute configurations of all of the stereogenic centers in **2** are the same as those found in **1**.

Minutissamide C (3) was obtained as a colorless, amorphous powder. The HRESIMS analysis (m/z 1210.6440 [M + Na]⁺) suggested the molecular formula of 3 to be $C_{55}H_{89}N_{13}O_{16}$. Detailed comparison of the NMR spectra indicated that compounds 1 and 3 share the same cyclic decapeptide core structure. The difference was the replacement of the dodecanoic acid residue found in 1 with an oxidized hexadecanoic acid residue in 3 (Table 2). The ¹H NMR spectrum of 3 displayed three additional methylene protons, H₂-12 ($\delta_{\rm H}$ 1.44), H₂-13 ($\delta_{\rm H}$ 2.38), and H₂-15 ($\delta_{\rm H}$ 2.40), as compared to 1. In addition to three methylene carbon signals ($\delta_{\rm C}$ 23.6, 41.9, and 35.4), a signal characteristic of a ketone ($\delta_{\rm C}$ 211.5) was also apparent in the ¹³C NMR spectrum. The molecular weight difference of 70 between 3 and 1 corresponded to three methylenes and one ketone, indicating the presence of a Hamoh (2-hydroxy-3-amino-4-methyl-14-oxohexadecanoic acid) moiety in 3. The location of the ketone group at C-14 in Hamoh was determined by the proton chemical shifts of H₂-13 ($\delta_{\rm H}$ 2.38) and H₂-15 ($\delta_{\rm H}$ 2.40) and a TOCSY correlation observed between H₂-15 ($\delta_{\rm H}$ 2.40) and H₃-16 (δ_H 0.90). An HMBC correlation from the terminal methyl protons (H₃-16, $\delta_{\rm H}$ 0.90) to the ketone carbon (C-14, $\delta_{\rm C}$ 211.5) further confirmed this location. The stereoconfiguration of 3 was assigned by comparison of spectroscopic data to those of 1. No significant difference (deviation <0.1 ppm) was found for the ¹³C chemical shifts of all of the stereogenic centers in 3 when compared to 1. Also, compound 3 displayed a similarly small positive specific rotation ($[\alpha]_{D}$ +3) to 1. Together, these data suggested the absolute configurations of the stereogenic centers in 3 to be identical to those of 1.

Minutissamide D (4) was also obtained as a colorless, amorphous powder. The HRESIMS spectrum of 4 displayed a major ion peak at 1212.6632 $[M + Na]^+$, suggesting a molecular formula of $C_{55}H_{91}N_{13}O_{16}$. The ¹H and ¹³C NMR spectra of 4 were almost identical to those of 3, except that the ketone in the

Hamoh moiety had been replaced by resonances for a carbinol moiety ($\delta_{\rm H}$ 3.28 and $\delta_{\rm C}$ 71.5). This, together with the molecular weight difference of 2 between 4 and 3, suggested that the ketone in Hamoh of 3 had been reduced to a hydroxy group in 4, forming a Hamhh residue (2-hydroxy-3-amino-4-methyl-14-hydroxyhexadecanoic acid). The structure of Hamhh was further supported by analysis of the COSY and HMBC spectra. Sequential COSY correlations between H₃-16 ($\delta_{\rm H}$ 0.83)/H₂-15 ($\delta_{\rm H}$ 1.16 and 1.35)/H-14 ($\delta_{\rm H}$ 3.28) and an HMBC correlation from H₃-16 $(\delta_{\rm H} 0.83)$ to C-14 $(\delta_{\rm C} 71.5)$ placed the hydroxy group at C-14. The absolute configurations of the amino acid residues and the three asymmetric centers in the Hamhh residue appeared to be the same as those found for 1 by comparison of ¹³C chemical shifts and specific rotation ($[\alpha]_D$ +2). Compound 4 possessed one more stereogenic hydroxy-bearing carbinol carbon (C-14) in the Hamhh residue. Mosher's ester analysis was attempted in an effort to assign the absolute configuration of this carbinol center. However, the presence of three hydroxy groups (one in Thr and two in Hamhh) resulted in a mixture of products and a very complex ¹H NMR spectrum of the Mosher's ester products of 4. Hydrolysis of 4 followed by isolation of the Hamhh residue would be necessary for Mosher's ester analysis, but was not attempted due to the limited amount of sample available. Thus, the absolute configuration of the C-13 carbinol carbon in Hamhh could not be assigned conclusively.

The structures of minutissamides A-D(1-4) were characterized by the presence of a β -amino acid residue, a 2-hydroxy-3amino-4-methyldodecanoic or -hexadecanoic acid, and three nonstandard amino acid residues (NMeAsn and two Dhbs). The β -amino acid residues were further modified by chlorination in 2, by oxidation to ketone in 3, or by hydroxylation in 4. Interestingly, the structures of minutissamides A-D(1-4) revealed some similarities to the cyclic decapeptides puwainaphycins A-E, isolated from a Hawaiian terrestrial Anabaena sp.9 The sequence of the five residues in the puwainaphycins, including Dhb, Val, β -amino acid unit, Pro, and NMeAsn, was conserved in the minutissamides. The other five amino acid residues, OMT (O-methyl-Thr), Gly, Gln, Thr-2 (or Val-2), and Thr-1, found in the puwainaphycins were replaced by Thr, Ala, Asn1, Dhb1, and Asn2 in the minutissamides. The core structure of the lipophilic β -amino acid residues, characterized by 2-hydroxy-3amino-4-methyl substitution, and the chlorination and oxidation patterns were also similar to those observed in the puwainaphycins.

Minutissamides A–D (1–4) were evaluated for their antiproliferative activity against the HT-29 human colon cancer cell line. Minutissamide A (1) displayed antiproliferative activity with an IC₅₀ value of 2.0 μ M, whereas lower activity was observed for minutissamides B–D (2–4) with IC₅₀ values of 20.0, 11.8, and 22.7 μ M, respectively. All four minutissamides have identical cyclic peptide cores and only differ in the lipophilic β -amino acid residue. For example, the only structural difference between minutissamides A (1) and B (2) was the presence of a chlorine atom at C-12 of this residue, but minutissamide A (1) was found to be 10-fold more active in the HT-29 assay. This suggests that the β -amino acid residue plays an important role in the antiproliferative activity of these compounds.

EXPERIMENTAL SECTION

General Experimental Procedures. Optical rotations were measured using a Perkin-Elmer 241 polarimeter. UV spectra were recorded on a Shimadzu UV spectrometer UV2401 and scanned from 190 to 360 nm. 1D and 2D NMR spectra were obtained on a Bruker Avance DRX 600 MHz NMR spectrometer with a 5 mm CPTXI Z-gradient probe and a Bruker Avance II 900 MHz NMR spectrometer with a 5 mm ATM CPTCI Z-gradient probe. ¹H and ¹³C NMR chemical shifts were referenced to the DMSO- d_6 solvent signals ($\delta_{\rm H}$ 2.50 and $\delta_{\rm C}$ 39.51, respectively). A mixing time of 60 ms was set for the TOCSY experiments and 200 ms for the T-ROESY experiment. The HMBC spectra were recorded with the ³ $J_{\rm C-H}$ set to 8 Hz, and the HSQC spectra were collected with the ¹ $J_{\rm C-H}$ set to 140 Hz. High-resolution ESI mass spectra were obtained using a Shimadzu IT-TOF LC mass spectrometer. Tandem mass analysis was performed using a Micromass Q-TOF LC mass spectrometer.

Biological Material. Anabaena minutissima was acquired from the Culture Collection of Algae at the University of Texas at Austin (UTEX 1613). The cyanobacterium was grown in 20 L flasks containing 18 L of inorganic media (Z media).¹³ Cultures were illuminated with fluorescent lamps at 1.03 klx. The temperature of the culture room was maintained at 22 °C. After 6–8 weeks, the biomass of cyanobacteria was harvested by centrifugation and then freeze-dried.

Extraction and Isolation. The freeze-dried biomass (5.2 g) from the total 54 L (3×18 L) culture was harvested and extracted with CH₂Cl₂-MeOH (1:1) and concentrated in vacuo to yield 1.1 g of the extract. The extract was fractionated using Diaion HP-20 with increasing amounts of IPA in H₂O to generate eight subfractions (0, 20, 40, 60, 70, 80, 90, 100% aqueous IPA). Fraction 3 (40% IPA) indicated the presence of four peptides by LC-MS. This fraction was further purified by Sephadex LH-20 column chromatography in MeOH. Fractions 7–9 containing the peptides were subjected to reversed-phased HPLC (Varian C₈ semipreparative column, 10 mm × 250 mm, 3 mL/min) eluting with a linear gradient from 60 to 80% aqueous MeOH for 45 min. Minutissamaides A–D (1–4) were eluted at 37.9 min (1, 1.4 mg), 31.6 min (2, 0.4 mg), 30.0 min (3, 1.0 mg), and 33.5 min (4, 0.7 mg), respectively.

Minutissamide A (**1**): colorless, amorphous powder; $[α]^{25}_{D}$ +4 (*c* 0.06, MeOH); UV (MeOH) $λ_{max}$ (log ε) 243 (3.45) nm; IR (neat) 3317 (br), 2925, 2861, 1681, 1635 (br), 1561, 1541 cm⁻¹; ¹H and ¹³C NMR (see Table 1); HRESIMS *m*/*z* 1140.6049 [M + Na]⁺ (calcd for C₅₁H₈₃N₁₃O₁₅Na, 1140.6029).

Minutissamide β (**2**): colorless, amorphous powder; $[\alpha]^{25}_{D}$ +5 (*c* 0.02, MeOH); UV (MeOH) λ_{max} (log ε) 230 (4.24) nm; IR (neat) 3315 (br), 2926, 2852, 1670, 1647 (br), 1545, 1516, 1459, 1402 cm⁻¹; ¹H and ¹³C NMR (see Table 1); HRESIMS *m*/*z* 1174.5680 [M + Na]⁺ (calcd for C₅₁H₈₃ClN₁₃O₁₅Na, 1174.5640).

Minutissamide C (**3**): colorless, amorphous powder; $[α]^{25}_{D}$ +3 (*c* 0.07, MeOH); UV (MeOH) $λ_{max}$ (log ε) 242 (2.62) nm; IR (neat) 3315 (br), 2926, 2857, 1664, 1630 (br), 1533, 1447, 1402 cm⁻¹; ¹H and ¹³C NMR (see Table 2); HRESIMS *m*/*z* 1210.6440 [M + Na]⁺ (calcd for C₅₅H₈₉N₁₃O₁₆Na, 1210.6448).

Minutissamide D (**4**): colorless, amorphous powder; $[α]^{25}_{D}$ +2 (*c* 0.05, MeOH); UV (MeOH) $λ_{max}$ (log ε) 240 (3.94) nm; IR (neat) 3320 (br), 2926, 2852, 1664, 1636 (br), 1539, 1453, 1379 cm⁻¹; ¹H and ¹³C NMR (see Table 2); HRESIMS *m*/*z* 1212.6632 [M + Na]⁺ (calcd for C₅₅H₉₁N₁₃O₁₆Na, 1212.6604).

Determination of Amino Acid Configurations. Approximately 0.3 mg of minutissamide A was hydrolyzed using 6 N HCl ($500 \,\mu$ L) in a high-pressure tube for 16 h at 110 °C. The hydrolysate was dried under vacuum and redissolved in H₂O. This procedure was repeated three times to completely remove the remaining HCl. For the derivatization with Marfey's reagent (FDAA), the hydrolysate (0.1 mg) or amino acid standard was dissolved in $50 \,\mu$ L of H₂O, and $20 \,\mu$ L of 1 N NaHCO₃ and 110 μ L of acetone were added. The reaction was initiated by adding 20 μ L of FDAA solution (10 mg/mL w/v in acetone), proceeded for 1 h at 40 °C, and was quenched by adding 20 μ L of 1 N HCl. The reactant was dried under vacuum and redissolved in CH₃CN for HPLC analysis. Chromatographic analysis of the FDAA

derivatives was performed on an Alltech C₁₈ reversed-phased column $(5 \ \mu m, 250 \times 4.6 \ mm)$ with the flow rate of 1.0 mL/min. Aqueous CH₃CN containing 0.1% formic acid (FA) was used as a mobile phase eluting with a linear gradient from 10 to 100% (A: H₂O containing 0.1% FA; B: CH₃CN containing 0.1% FA; gradient conditions: 0 min 10% B, 5 min 10% B, 35 min 20% B, 65 min 30% B, 70 min 60% B, 72 min 100% B, and 80 min 100% B). The absolute configurations of the amino acids were assigned by comparing the retention times with those of the corresponding amino acid standards. The retention times of the FDAA derivatives of the authentic amino acid standards were identified at 30.70 (L-Thr and N-Me-D-Asp), 32.30 (L-Asp), 36.17 (N-Me-L-Asp), 36.94 (D-Asp), 41.06 (L-Ala), 42.54 (D-Thr), 44.57 (L-Pro), 49.84 (D-Pro), 51.76 (D-Ala), 59.09 (L-Val), and 71.45 (D-Val) min. The FDAA derivatives of the amino acids from the hydrolysate of 1 showed the peaks at 32.30, 36.17, 44.57, 51.76, and 59.09 min, corresponding to L-Asp, N-Me-L-Asp, L-Pro, and L-Val, respectively. The peak at 30.70 min was identified as L-Thr by LC-MS analysis $(m/z 372, [M + H]^+)$.

Advanced Marfey's Analysis. L- and D-FDLA (1-fluoro-2,4dinitrophenyl-5-leucinamide) was synthesized using the protocol published for Marfey's analysis.²¹ The acid hydrolysate (0.2 mg) was separated into two equal portions for derivatization with either L-FDLA or DL-FDLA. Each portion was dissolved in 50 μ L of H₂O and then mixed with 20 μ L of 1 N NaHCO₃ and 110 μ L of acetone. Finally, 20 μ L of L-FDLA or DL-FDLA (10 mg/mL w/v in acetone) was added, and the mixtures were heated to 40 °C for 1 h. The reaction mixtures were cooled to room temperature, and $20 \,\mu\text{L}$ of 1 N HCl was added to quench the reaction. After drying, the FDLA derivatives were dissolved in CH₃CN for LC-MS analysis. The chromatograms of L-FDLA and DL-FDLA derivatives were compared for the assignment of the absolute configuration of Hamd in 1. LC-MS analysis was performed using a Varian reverse-phased C18 column $(250 \times 2.0 \text{ mm})$ at the flow rate of 0.4 mL/min. The linear gradient eluting from 20 to 75% aqueous CH₃CN (0.1% FA) for 50 min was used as a mobile phase, and ESI was used as the ionization method. Two peaks corresponding to the L- and D-FDLA derivatives of Hamd were observed at 43.33 and 48.84 min $(m/z 538, [M - H]^{-})$, respectively. The L-FDLA derivative of Hamd gave one peak at 48.84 min.

HT-29 Antiproliferative Assay. Antiproliferative activity against the HT-29 cancer cell line was performed according to established protocols.²²

ASSOCIATED CONTENT

Supporting Information. Complete NMR data tables of 2-4; HPLC chromatograms of Marfey's and the advanced Marfey's derivatives of 1; ¹H, COSY, TOCSY, HSQC, and HMBC spectra of 1-4; DEPTQ spectra of 1, 3, and 4; ROESY spectrum of 1; Q-TOF Tandem MS spectrum of 1. This information is available free of charge via the Internet at http://pubs.acs.org.

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REFERENCES

(1) Tan, L. T. Phytochemistry 2007, 68, 954-979.

(2) Wagoner, R. M. V.; Drummond, A. K.; Wright, J. L. C. Adv. Appl. Microbiol. 2007, 61, 89–217.

(3) Harada, K. I. Chem. Pharm. Bull. 2004, 52, 889-899.

(4) Singh, S.; Kate, B. N.; Banerjee, U. C. Crit. Rev. Biotechnol. 2005, 25, 73–95.

(5) Burja, A. M.; Banaigs, B.; Abou-Mansour, E.; Burgess, J. G.; Wright, P. C. *Tetrahedron* **2001**, *57*, 9347–9377.

(6) MacMillan, J. B.; Ernst-Russell, M. A.; Ropp, J. S.; Molinski, T. F. J. Org. Chem. 2002, 67, 8210–8215.

(7) Frankmölle, W. P.; Knübel, G. K.; Moore, R. E.; Patterson, G. M. L. J. Antibiot. 1992, 45, 1458–1466.

(8) Moore, R. E.; Bornemann, V.; Niemczura, W. P.; Gregson, J. M.; Chen, J. L.; Norton, T. R.; Patterson, G. M. L.; Helms, G. L. *J. Am. Chem. Soc.* **1989**, *111*, 6128–6132.

(9) Gregson, J. M.; Chen, J. L.; Patterson, G. M. L.; Moore, R. E. Tetrahedron 1992, 48, 3727–3734.

(10) Gerwick, W. H.; Jiang, Z. D.; Agarwal, S. K.; Farmer, B. T. *Tetrahedron* **1992**, 48, 2313–2324.

(11) Bonnard, I.; Rolland, M.; Salmon, J. M.; Debiton, E.; Barthomeuf, C.; Banaigs, B. J. Med. Chem. **2007**, *50*, 1266–1279.

(12) Welker, M.; Döhren, H. FEMS Mricrobiol. Rev. 2006, 30, 530-563.

(13) Falch, B. S.; Konig, G. M.; Wright, A. D.; Sticher, O.; Angerhofer,
 C. K.; Pezzuto, J. M.; Bachmann, H. *Planta Med.* **1995**, *61*, 321–328.

 (14) Claridge, T. D. W.; Pérez-Victoria, I. Org. Biomol. Chem. 2003, 1, 3632–3634.

(15) Eckart, K.; Schwarz, H.; Tomer, K. B.; Gross, M. L. J. Am. Chem. Soc. 1985, 107, 6765–6769.

(16) Bhushan, R.; Brückner, H. Amino Acids 2004, 27, 231-247.

(17) Ding, K. Magn. Reson. Chem. 2000, 38, 321-323.

(18) Fujii, K.; Ikai, Y.; Oka, H.; Suzuki, M.; Harada, K. Anal. Chem. 1997, 69, 5146–5151.

(19) Harada, K.; Fujii, K.; Hayashi, K.; Suzuki, M. *Tetrahedron Lett.* **1996**, 37, 3001–3004.

(20) Fujii, K.; Shimoya, T.; Ikai, Y.; Oka, H.; Harada, K. *Tetrahedron Lett.* **1998**, 39, 2579–2582.

(21) Marfey, P. Carlsberg Res. Commun. 1984, 49, 591-596.

(22) Seo, E.-K.; Kim, N.-C.; Mi, Q.; Chai, H.; Wall, M. E.; Wani,

M. C.; Navarro, H. A.; Burgess, J. P.; Graham, J. G.; Cabieses, F.; Tan, G. T.; Farnsworth, N. R.; Pezzuto, J. M.; Kinghorn, A. D. *J. Nat. Prod.* **2001**, *64*, 1483–1485.